

**A Novel Approach to the Stereocontrolled Synthesis of Steroid Side Chains
Including the CD-Ring System: The First Total Synthesis of
(+)-8 α -(Phenylsulfonyl)-des-AB-cholestane and Its Efficient Conversion
into Grundmann's Ketone and Vitamin D₃[†]**

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A new stereocontrolled approach to steroid CD-ring system including side chain starting from an optically active indenedione **6** is described. The application of this finding allows for the asymmetric synthesis of des-AB-cholestane **3** and 8 α -(phenylsulfonyl)-des-AB-cholestane **37**; from the latter Grundmann's ketone **25** and vitamin D₃ (**27**) were synthesized efficiently.

In the total synthesis of biologically important steroidal compounds, the stereocontrolled construction of the CD-ring system and side chain is a major problem, and a number of investigations into a solution to this problem have been reported to date. Namely, catalytic hydrogenation¹ of $\Delta^{17(20)}$ -, $\Delta^{20(22)}$ -, or $\Delta^{20(21)}$ -olefins, [3,3]-sigmatropic^{2a-c} and [2,3]-Wittig^{2h,i} rearrangements, substitution reactions of π -allylpalladium intermediates,³ boron intermediates,⁴ and allyl alcoholic esters,⁵ addition reactions of allylic epoxides⁶ and enones,⁷ ene reactions,⁸ cycloaddition reactions,⁹ and radical cyclization reactions¹⁰ have been used as key stereodirecting processes. The use of the conformational rigidity of bicyclo[2.2.1]heptane derivatives¹¹ has also led to an interesting solution to this problem. Although various types of reaction exemplified above have been used to explore an efficient route for stereoselective construction of steroid side chain including CD-ring system, a direct and practical process still remains to be developed for bringing totally synthesized, biologically important steroidal compounds to market. Accordingly, a program for the total synthesis of steroidal CD-ring compounds having appropriate side chains in optically pure form was undertaken. This paper describes the stereocontrolled synthesis of (-)-des-AB-8-oxacholest-14-en-9-one (**3**) and the first total synthesis of (+)-8 α -(phenylsulfonyl)-des-AB-cholestane **37**, both in optically pure forms. Compound **37** was then converted into Grundmann's ketone (**25**) and vitamin D₃ (**27**).

Synthetic Plan for (-)-Des-AB-8-oxacholest-14-en-9-one (3). The first target compound was (-)-des-AB-8-oxacholest-14-en-9-one (**3**) because des-AB-cholestan-9-one (**1**) and its ester derivative (**2**), both of which had been prepared from **3**,^{7b} were known to be flexible intermediates for annelative elaboration into cholestane-like steroids or for transformation into members of the vitamin D series.¹² The starting material should be readily available in large quantities and also optically pure form for our concept to be viable, and the indenedione **6**¹³ seemed to be an ideal candidate for this purpose. The overall plan is summarized in Scheme I¹⁶ wherein the selective bond scission in cyclohexane ring of the indenedione **5**, which could be prepared from **6**, would give the cyclopentane **4a** which is redrawn as **4** for convenience. Reduction of the carboxylic acid to a methyl group, homologation of the X portion, and then cyclization of **4** would provide the initial target com-

pound **3**. The most significant feature of this plan is that the chiral centers at C(13), C(17), and C(20) (steroid numbering) are incorporated in compound **5** as the most stable isomeric form.

Results. The reaction¹⁷ of the optically pure (-)-indenedione **6** with paraformaldehyde and thiophenol in triethanolamine afforded the sulfide **7** (Scheme II), which was then oxidized (MCPBA) to give the sulfone **8**, in 39%

(1) (a) Piatak, D. M.; Wicha, J. *Chem. Rev.* **1978**, *78*, 199. (b) Kametani, T.; Tsubuki, M.; Nemoto, H. *Tetrahedron Lett.* **1981**, *22*, 2373; (c) *J. Chem. Soc., Perkin Trans. 1* **1981**, 3077.

(2) (a) Chapleo, C. B.; Hallett, P.; Lythgoe, B.; Waterhouse, I.; Wright, P. W. *J. Chem. Soc., Perkin Trans. 1* **1977**, 1211. (b) Tanabe, M.; Hayashi, K. *J. Am. Chem. Soc.* **1980**, *102*, 862. (c) Koreeda, M.; Tanaka, Y.; Schwartz, A. *J. Org. Chem.* **1980**, *45*, 1172. (d) Takahashi, T.; Yamada, H.; Tsuji, J. *J. Am. Chem. Soc.* **1981**, *103*, 5259; (e) *Tetrahedron Lett.* **1982**, *23*, 233. (f) Stork, G.; Atwal, K. S. *Ibid.* **1982**, *23*, 2073. (g) Ziegler, F. E.; Lim, H. *J. Org. Chem.* **1984**, *49*, 3278. (h) Castedo, L.; Grandja, J. R.; Mourino, A. *Tetrahedron Lett.* **1985**, *26*, 4719. (i) Mikami, K.; Kawamoto, K.; Nakai, T. *Ibid.* **1985**, *26*, 5799.

(3) (a) Trost, B. M.; Verhoeven, T. R. *J. Am. Chem. Soc.* **1978**, *100*, 3435. (b) Temple, J. S.; Schwartz, J. *Ibid.* **1980**, *102*, 7381. (c) Riediker, M.; Schwartz, J. *Tetrahedron Lett.* **1981**, *22*, 4655.

(4) (a) Midland, M. M.; Kwon, Y. C. *J. Org. Chem.* **1981**, *46*, 229; (b) *Tetrahedron Lett.* **1982**, *23*, 2077.

(5) Schuff, N. R.; Trost, B. M. *J. Org. Chem.* **1983**, *48*, 1404.

(6) (a) Mariano, J. P.; Abe, H. *J. Am. Chem. Soc.* **1981**, *103*, 2907. (b) Takahashi, T.; Ootake, A.; Tsuji, J. *Tetrahedron Lett.* **1984**, *25*, 1921.

(7) (a) Stork, G.; Shiner, C. S.; Winkler, J. D. *J. Am. Chem. Soc.* **1982**, *104*, 310, 3767. (b) Ziegler, F. E.; Mencil, J. J. *Tetrahedron Lett.* **1983**, *24*, 1859. (c) Takahashi, T.; Okumoto, H.; Tsuji, J.; Harada, N. *J. Org. Chem.* **1984**, *49*, 948. (d) Takahashi, T.; Okumoto, H.; Tsuji, J. *Tetrahedron Lett.* **1984**, *25*, 1925. (e) Fukuzaki, K.; Nakamura, E.; Kuwajima, I. *Ibid.* **1984**, *25*, 3591.

(8) (a) Dauben, W. G.; Brookhart, T. J. *J. Am. Chem. Soc.* **1981**, *103*, 237. (b) Batcho, A. D.; Berger, D. E.; Uskoković, M. R. *Ibid.* **1981**, *103*, 1293.

(c) Dauben, W. G.; Brookhart, T. J. *J. Org. Chem.* **1982**, *47*, 3921. (d) Wovkulich, P. M.; Baggiolini, E. G.; Hennessy, B. M.; Uskoković, M. R.; Mayer, E.; Norman, A. W. *Ibid.* **1983**, *48*, 4433.

(9) (a) Desmaele, D.; Ficini, J.; Guingant, A.; Kahn, Ph. *Tetrahedron Lett.* **1983**, *24*, 3079. (b) Wilson, S. R.; Haque, M. S. *Ibid.* **1984**, *25*, 3147.

(c) Nemoto, H.; Nagai, M.; Fukumoto, K.; Kametani, T. *Ibid.* **1985**, *26*, 4613; (d) *J. Org. Chem.* **1985**, *50*, 2764; (e) *Tetrahedron* **1985**, *41*, 2361.

(10) Stork, G.; Kahn, M. J. *J. Am. Chem. Soc.* **1985**, *107*, 500.

(11) (a) Stevens, R. V.; Gaeta, F. C. A. *J. Am. Chem. Soc.* **1977**, *99*, 6105. (b) Trost, B. M.; Bernstein, P. R.; Funschilling, P. C. *Ibid.* **1979**, *101*, 4378. (c) Grieco, P. A.; Takigawa, T.; Moore, D. R. *Ibid.* **1979**, *101*, 4380. (d) Stevens, R. V.; Gaeta, F. C. A.; Lawrence, D. S. *Ibid.* **1983**, *105*, 7713. (e) Hutchinson, J. H.; Money, T. *Tetrahedron Lett.* **1985**, *26*, 1819.

(12) Baggiolini, E. R.; Iacobelli, J. A.; Hennessy, B. M.; Uskoković, M. *J. Am. Chem. Soc.* **1982**, *104*, 2945.

(13) This compound has been synthesized in high chemical and optical yield by G. Ohloff et al.¹⁴ by following the same procedure as for the enantiomer of the compound **6** using (+)-proline instead of (-)-proline as a catalyst which had been developed originally by Z. G. Hajos et al.¹⁵

(14) Ohloff, G.; Maurer, B.; Winter, B.; Giersch, W. *Helv. Chim. Acta* **1983**, *66*, 192.

(15) Hajos, Z. G.; Parrish, D. R. *J. Org. Chem.* **1974**, *39*, 1615.

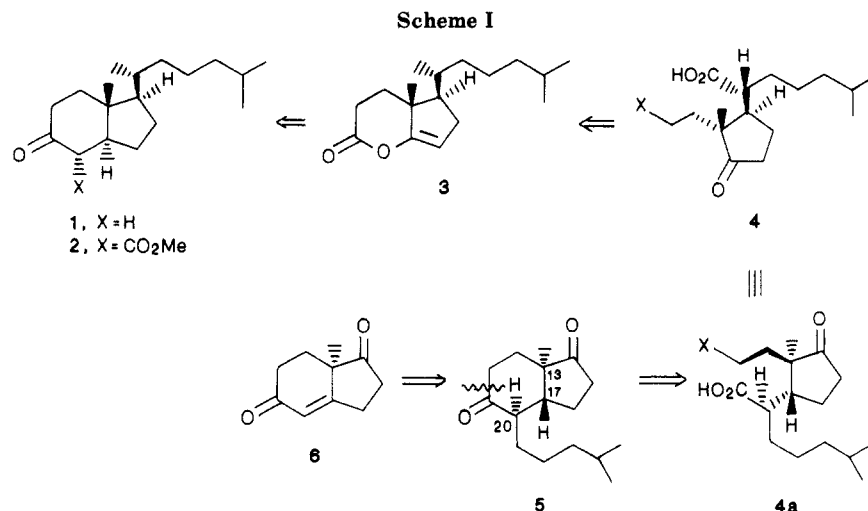
(16) All structures appeared on this manuscript show absolute configurations unless otherwise stated.

(17) For related references, see: Kirk, D. N.; Petrow, V. *J. Chem. Soc.* **1962**, 1091. See also: Sauer, G.; Eder, U.; Haffer, G.; Neef, G.; Wiechert, R. *Angew. Chem., Int. Ed. Engl.* **1975**, *14*, 417.

[†]The material presented in this manuscript has been preceded by a preliminary communication published in the following: *Chem. Lett.* **1985**, 259.

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overall yield from 6. Base-catalyzed elimination of the trans sulfone 9, obtained in 71% yield by catalytic hydrogenation (H₂, Pd-C) of 8, furnished the methylene ketone 10¹⁸ in 98% yield. Next, the isoamyl group was added in a 1,4-fashion to 10 by using isoamylmagnesium bromide in THF under CuI/BF₃·Et₂O catalysis²⁰ to give a 73% yield of the dione 5, *m/z* 250 (M⁺). The enol acetylation (isopropenyl acetate, *p*-TsOH) of 5 afforded the dienol diacetate 11, *m/z* 334 (M⁺), and the enol monoacetate 12, *m/z* 292 (M⁺), in a ratio of 2:1 in 94% yield, and the diacetate 11 was found to be hydrolyzed (LiOH, aqueous THF) selectively to give the monoacetate 12 in 86% yield. The keto acid 13 which was obtained by the ozonolysis (O₃, CH₂Cl₂; Me₂S) of 12, followed by a hydrolysis (LiOH, aqueous THF) of the resulting acid anhydride, was converted as usual (HOCH₂CH₂OH, camphorsulfonic acid) into the acetal carboxylic acid 14, in 70% yield. The transformation of 14 into the methyl derivative 18 was straightforward as described below. The carboxylic acid 14 was subjected to reduction (LiAlH₄) to the diol 15, *m/z* 314 (M⁺), which was converted (*p*-TsCl, TEA, DMAP) into the monotosylate 16, *m/z* 468 (M⁺), in 62% overall yield from 14. The compound 17, *m/z* 512 (M⁺), prepared in 84% yield by the protection (MOMCl, Hünig base) of 16, was then subjected to reduction (LiAlH₄) to give the methyl derivative 18, *m/z* 342 (M⁺), in 92% yield. Selective deprotection (10% HCl, acetone) of the acetal group of 18 afforded the aldehyde 19, in 92% yield, and then this was transformed into the ester 21, in 72% yield, via the enol ether 20, *m/z* 326 (M⁺), by a Wittig reaction²¹ (MeOCH₂POPh₂, LDA; then NaH) followed by oxidation²² (PCC) of the resulting enol ether 20. The lactone 22, *m/z* 266 (M⁺), [prepared by deprotection (10% HCl, acetone) of the ester 21] was hydrolyzed (LiOH, aqueous THF) and then oxidized (Jones' reagent) to afford

in 81% overall yield from 21 the keto carboxylic acid 24, which was identical by IR (neat) and ¹H NMR (CDCl₃, 300 MHz) spectral comparison with an authentic sample.^{7b} The keto carboxylic acid 24 thus obtained was transformed into the enol lactone 3, [α]_D - 32.5°, according to the known procedure.^{7b}

Thus, the synthetic strategy described above provides a new method for stereoselective construction of the steroid CD-ring system and side chain.

Synthesis of Vitamin D₃. Because of the important therapeutic value of vitamin D in treating disorders of calcium and phosphorus metabolism, considerable attention has been directed toward the chemistry of vitamin D leading to the discovery of vitamin D derivatives possessing high biological activity and increased activity.^{8b,d,11b,c,d,12,24} In Scheme III is illustrated the synthetic approach to these compounds by Julia's olefin synthesis, namely the reductive elimination of β-alkoxy sulfone 27, (prepared by condensation of sulfone 25 with aldehyde 26), which gives stereoselectively vitamin D₄ (28).^{25a} This has provided a limited entry to the vitamin D series mainly because of the difficulty in obtaining the starting sulfone 25 (synthesis of which requires several steps from the corresponding 8-keto derivative). These facts and the desirable feature that the trans-fused sulfone 25 does not epimerize the corresponding cis-fused sulfone under basic conditions in which Grundmann's ketone could be epimerized to the corresponding cis-isomer stimulated us to develop a direct and efficient method for synthesis of 8α-(phenylsulfonyl)-des-AB-cholestane (30) and its application to the synthesis of vitamin D₃.

Synthetic Plan. The overall synthetic plan of 30 summarized in Scheme IV is inherently the same as for that of 3 in Scheme I except for the stereoselective hydrogenation of 34, which could be prepared from 5, giving the keto sulfone 33 and the intramolecular alkylation of 31, which could be obtained from 33 via 32a (32) by bond scission of 33, followed by reduction of the carboxyl group to a methyl group and homologation of the X portion of the resulting compound 32a (32).

Certainly the most provocative feature of this plan is that all the chiral centers at C(13), C(14), C(17), and C(20) (steroid numbering) required for the synthesis of 30 are incorporated in compound 33 as the most stable isomeric form since the chirality at C(8) could be generated selec-

(18) This type of compound had been prepared previously through different types of reaction sequences.¹⁹

(19) Hajos, Z. G.; Parrish, D. R. *J. Org. Chem.* 1973, 38, 3244.

(20) Yamamoto, Y.; Yamamoto, S.; Yatagai, H.; Ishihara, Y.; Maruyama, K. *J. Org. Chem.* 1982, 47, 119.

(21) Earnshaw, C.; Wallis, C. J.; Warren, S. *J. Chem. Soc., Perkin Trans. 1* 1979, 3099.

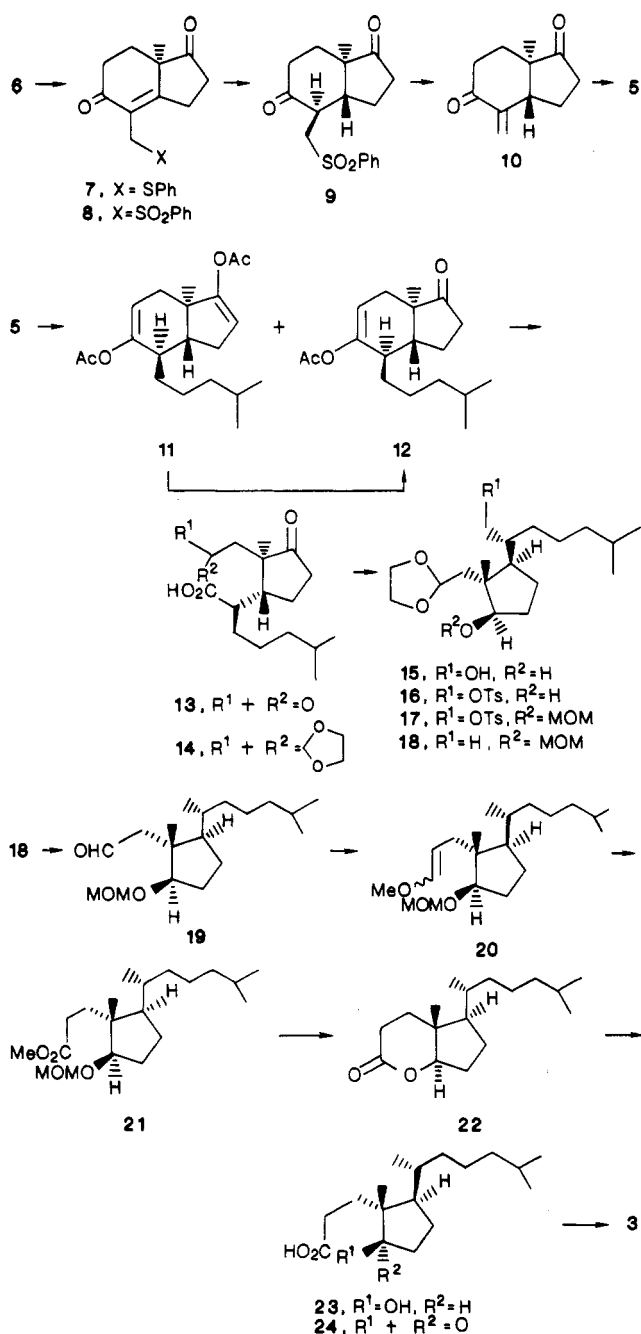
(22) Piancatelli, G.; Scettri, A.; D'Auria, M. *Tetrahedron Lett.* 1977, 3483.

(23) (a) DeLuca, H. F. *Fed. Proc., Fed. Am. Soc. Exp. Biol.* 1974, 33, 2211. (b) Wing, R. M.; Okamura, W. H.; Pirio, M. R.; Sine, S. M.; Norman, A. W. *Science (Washington, D.C.)* 1974, 189, 939. (c) Lawson, D. E. M.; Fraser, D. R.; Kodicek, E.; Morris, H. R.; Williams, D. H. *Nature (London)* 1971, 230, 228. (d) DeLuca, H. F.; Schnoes, H. K. *Ann. Rev. Biochem.* 1976, 45, 631. (e) Wichmann, J. K.; DeLuca, H. F.; Schnoes, H. K.; Horst, R. L.; Shepard, R. M.; Jorgensen, N. A. *Biochemistry* 1979, 18, 4775.

(24) Lythgoe, B. *Chem. Soc. Rev.* 1980, 9, 449 and references cited therein.

(25) (a) Kocienski, P. J.; Lythgoe, B.; Ruston, S. *J. Chem. Soc., Perkin Trans. 1* 1979, 1290. (b) Green, M. *J. Chem. Soc.* 1963, 1324.

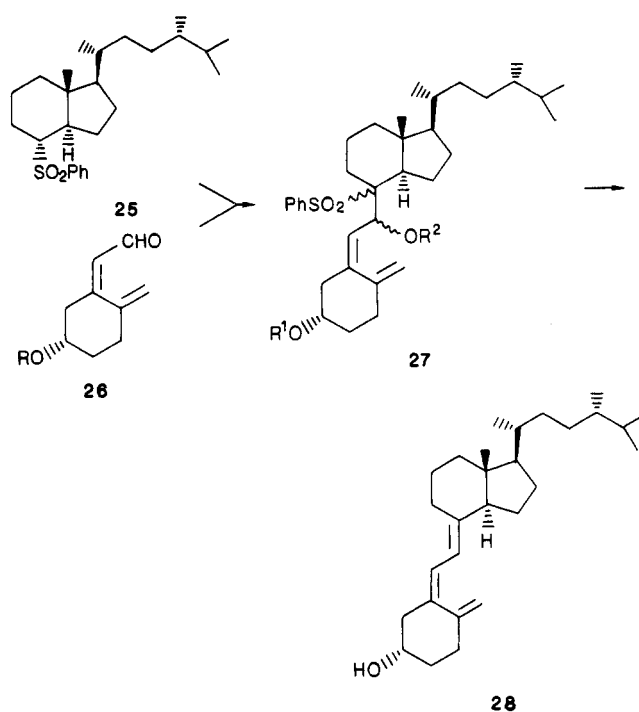
Scheme II



tively under thermodynamically controlled conditions. Finally, by following the same sequence of reactions shown in Scheme III, the sulfone 30 could be converted into vitamin D₃ (29).

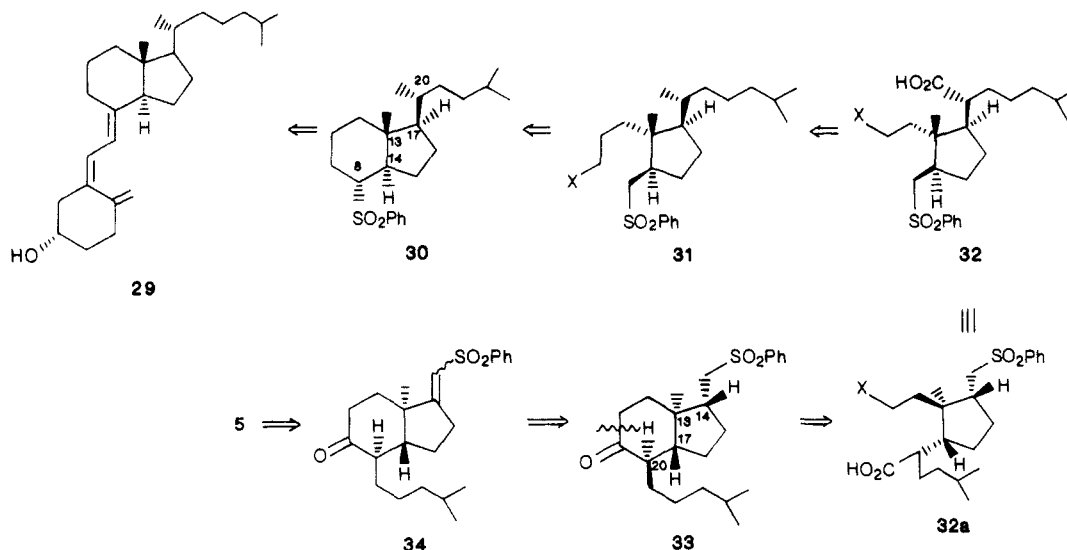
Results. As shown in Scheme V, the monoketal (35), obtained in 85% yield by a selective ketalization (HOCH₂CH₂OH, *p*-TsOH) of the diketone 5, was subjected to a Wittig reaction^{26b} (PhSCH₂PO(OEt)₂, NaH) affording the vinyl sulfide 36, *m/z* 400 (M⁺), in 95% yield as a mixture of *E/Z* isomers. Oxidation (MCPBA) gave the corresponding vinyl sulfone 37 as a separable mixture of *Z* (37a) and *E* (37b) isomers, *m/z* 432 (M⁺), in 80% total yield. Hydrogenation (H₂, Pd-C, 10% HCl, MeOH) of 37 afforded stereoselectively in 72% yield the keto sulfone 33, *m/z* 390 (M⁺), followed by enol acetylation (isopropenyl acetate, *p*-TsOH) to furnish the enol acetate 38, *m/z* 432 (M⁺), in 98% yield. Successive reaction of 38, namely, ozonolysis (O₃, CH₂Cl₂; Me₂S), hydrolysis (LiOH, aqueous THF), and then acetalization (HOCH₂CH₂OH, camphor-

Scheme III



sulfonic acid) formed the acid acetal (39), *m/z* 466 (M⁺), in 73% overall yield. The alcohol 40, *m/z* 452 (M⁺), obtained in 72% yield by reduction (LiAlH₄) of 39, was then converted into the methyl derivative 42, *m/z* 436 (M⁺), in 95% yield via tosylation (*p*-TsCl, pyridine, DMAP) followed by reduction (LiAlH₄). The final stage of this synthesis involved a one-carbon homologation and intramolecular alkylation. To accomplish this, acetal 42 was hydrolyzed (10% HCl, acetone) to the aldehyde 43, *m/z* 392 (M⁺), in 95% yield. The intramolecular alkylation was achieved by two different routes. Thus, in the first sequence epoxide 44 obtained in 31% yield by the epoxidation (Me₃S⁺I⁻, *n*-BuLi) of aldehyde 43, was subjected to the intramolecular cyclization (LDA, THF) giving the alcohols 45a and 46a in 30% and 50% yields, respectively. In the ¹H NMR spectrum, the signals observed at 3.55–3.76 ppm as multiplet due to methylene protons of hydroxymethylene moiety in the compound 45a were shifted to 3.80–4.21 ppm in its acetoxy derivative 45b. The α -configuration of a phenylsulfonyl group at C-8 in 46a was deduced from the coupling constants (3.02 ppm, d,d,d, *J* = 12, 12, 4 Hz) of C-8H in the NMR spectrum. This was eventually confirmed by conversion of 46a into the target compound 30, [α]_D +0.06°, in 70% overall yield via the mesylate 46b and olefins 47 by successive treatment (MsCl, pyridine; LiBr, Li₂CO₃; H₂, Pd-C). Because of the low yield of the epoxidation step and the lack of regioselectivity in the intramolecular cyclization of the epoxide 44 thus obtained, a second pathway to 30 was investigated. Thus, the aldehyde 43 was subjected to Peterson's olefin synthesis (Me₃SiCH₂MgCl; NaH), affording the olefin 48, *m/z* 390 (M⁺), in 81% yield. The alcohol 49, *m/z* 408 (M⁺), obtained in 92% yield by hydroboration-oxidation (BF₃·SMe₂; 30% H₂O₂, NaOH) of 47 was first mesylated (MsCl, pyridine) to give mesylate 50 in 98% yield and then subjected to intramolecular alkylation (LDA) to furnish the target compound 30 in 84% yield. This was identical with the authentic sample prepared above in all aspects including optical rotation and was converted by oxidation²⁶

Scheme IV



[oxodiperoxymolybdenum-pyridine-hexamethylphosphoric triamide (MoOPH),²⁷ LDA] in 62% yield into Grundmann's ketone 51 which was identical with an authentic sample.²⁸ By this conversion to 51, it was confirmed that all the chiral centers except C-8 of the compound 30 were introduced correctly. To this end, the compound 30 was converted into vitamin D₃ as follows. The sulfone was metalated (LDA) and then condensed with the ring A component 53, obtained by oxidation (MnO₂) of the corresponding allyl alcohol 52.²⁹ Treatment of the reaction mixture with acetyl chloride gave a mixture of diastereoisomeric β -acetoxysulfones 54. This mixture was reduced (5% Na-Hg) to the triene 55, whose desilylation (*n*-Bu₄N⁺F⁻) gave vitamin D₃ (29) in 51% overall yield. The 3,5-dinitrobenzoate 56 of the synthetic vitamin D₃ (mp 129–130 °C, $[\alpha]_D^{20} +95.9^\circ$) was identical with the authentic vitamin D₃ 3,5-dinitrobenzoate (lit.³⁰ mp 128–129 °C, lit.³¹ $[\alpha]_D^{20} +97^\circ$) in mixed melting point test and spectral (IR and ¹H NMR) comparisons.

A direct and efficient method for the synthesis of 8 α -(phenylsulfonyl)-des-AB-cholestane (30) in an optically pure form has been disclosed. We have also demonstrated the versatility of 30 in the synthesis of vitamin D analogues by converting it into Grundmann's ketone (51) and vitamin D₃ (29). The methodology described above opens up many fascinating possibilities in the asymmetric synthesis of biologically important steroids.

Experimental Section

All melting points were uncorrected. Infrared (IR) spectra were recorded on a Hitachi 260-10 spectrophotometer. Nuclear magnetic resonance (NMR) spectra were measured on JEOL-PMX-60 and JEOL-PS-100 spectrometer. Chemical shifts were reported as δ values relative to internal tetramethylsilane (Me₄Si). Mass spectra were taken on Hitachi RMU-7, Hitachi M-52G, and JEOL-JMS-OISG-2 spectrometers. All optical rotations were measured in chloroform solution on a JASCO DIP-4 polarimeter using a 1-dm cell.

(-)-(7aR)-4-[(Phenylthio)methyl]-5,6,7,7a-tetrahydro-7a-methylindan-1,5-dione (7). To a stirred solution of indenedione¹³ 6 (50 g, 304.9 mmol) in triethanolamine (100 mL) was added 80%

paraformaldehyde (14 g) at 60 °C, and stirring was continued for 20 min at the same temperature. To it was added dropwise thiophenol (34.4 mL), and the mixture was stirred for additional 24 h at the same temperature. The reaction mixture was then poured into water (300 mL) and extracted with ether (300 mL \times 3). The combined extracts were washed with saturated NaCl solution and dried (Na₂SO₄). Removal of the solvent afforded a crude product which was chromatographed on silica gel (300 g) using hexane-ethyl acetate (4:1 v/v) as an eluant to give 38.4 g (44%) of the sulfide 7 as colorless prisms: mp 101–102 °C (MeOH); IR (CHCl₃) 1750, 1680 cm⁻¹; NMR (CDCl₃) δ 1.15 (3 H, s, CH₃), 3.63 (2 H, s, CH₂S), 7.00–7.30 (5 H, m, Ar H); mass spectrum, *m/z* 286 (M⁺); $[\alpha]_D^{20} -227^\circ$ (*c* 0.47). Anal. Calcd for C₁₇H₁₈O₂S: C, 71.29; H, 6.34; S, 11.20. Found: C, 71.13; H, 6.21; S, 11.04.

(-)-(7aR)-4-[(Phenylsulfonyl)methyl]-5,6,7,7a-tetrahydro-7a-methylindan-1,5-dione (8). To a stirred solution of the sulfide 7 (728 mg, 2.55 mmol) in CH₂Cl₂ (20 mL) was added in portions 70% *m*-chloroperbenzoic acid (1.26 g, 5.1 mmol) at room temperature. After stirring for 3 h at the same temperature, the reaction mixture was washed with saturated NaHCO₃ and NaCl solutions, and dried (Na₂SO₄). The residue resulting from evaporation of the solvent was chromatographed on silica gel (20 g) using hexane-ethyl acetate (3:2 v/v) as an eluant to afford 710 mg (88%) of the sulfone 8 as colorless needles: mp 164.5–165 °C (MeOH); IR (CHCl₃) 1750, 1680 cm⁻¹; NMR (CDCl₃) δ 1.33 (3 H, s, CH₃), 4.05, 4.35 (2 H, each d, *J* = 14 Hz, CH₂SO₂), 7.33–7.97 (5 H, m, Ar H); mass spectrum, *m/z* 318 (M⁺); $[\alpha]_D^{20} -212^\circ$ (*c* 0.33). Anal. Calcd for C₁₇H₁₈O₄S: C, 64.13; H, 5.70; S, 10.07. Found: C, 63.85; H, 5.69; S, 10.45.

(-)-(3aR,4R,7aR)-4-[(Phenylsulfonyl)methyl]-3a,4,5,6,7,7a-hexahydro-7a-methylindan-1,5-dione (9). A mixture of 8 (6.2 g, 19.5 mmol), 5% palladium on carbon (700 mg), AcOH (45 mL), 10% HCl (2.3 mL), and MeOH (160 mL) was stirred under an atmosphere of hydrogen at room temperature. Hydrogenation was completed within 4 h. The solution was then filtered to remove the catalyst, and the catalyst was washed with MeOH. The filtrate and washing were combined and evaporated to leave a residue, which was dissolved in benzene (200 mL). The organic layer was washed with water and dried (Na₂SO₄). The removal of the solvent afforded 4.42 g (71%) of 9 as colorless prisms: mp 118–119 °C (MeOH); IR (CHCl₃) 1740, 1720 cm⁻¹; NMR (CDCl₃) δ 1.23 (3 H, s, CH₃), 4.00 (2 H, dd, *J* = 5 Hz and 13 Hz, CH₂SO₂), 7.45–8.00 (5 H, m, Ar H); mass spectrum, *m/z* 320 (M⁺); $[\alpha]_D^{20} -96.7^\circ$ (*c* 0.24). Anal. Calcd for C₁₇H₂₀O₄S: C, 63.72; H, 6.29; S, 10.01. Found: C, 63.35; H, 6.16; S, 10.10.

(-)-(3aR,7aR)-4-Methylene-3a,4,5,6,7,7a-hexahydro-7a-methylindan-1,5-dione (10). To stirred solution of 9 (1 g, 3.1 mmol) in benzene (40 mL) was added dropwise 1,8-diazabicyclo[5.4.0]undec-7-ene (0.5 g) at room temperature, and stirring was continued for 30 min at the same temperature. The reaction

(27) Vedejs, E.; Engler, D. A.; Telschow, J. E. *J. Org. Chem.* 1978, 43, 188.

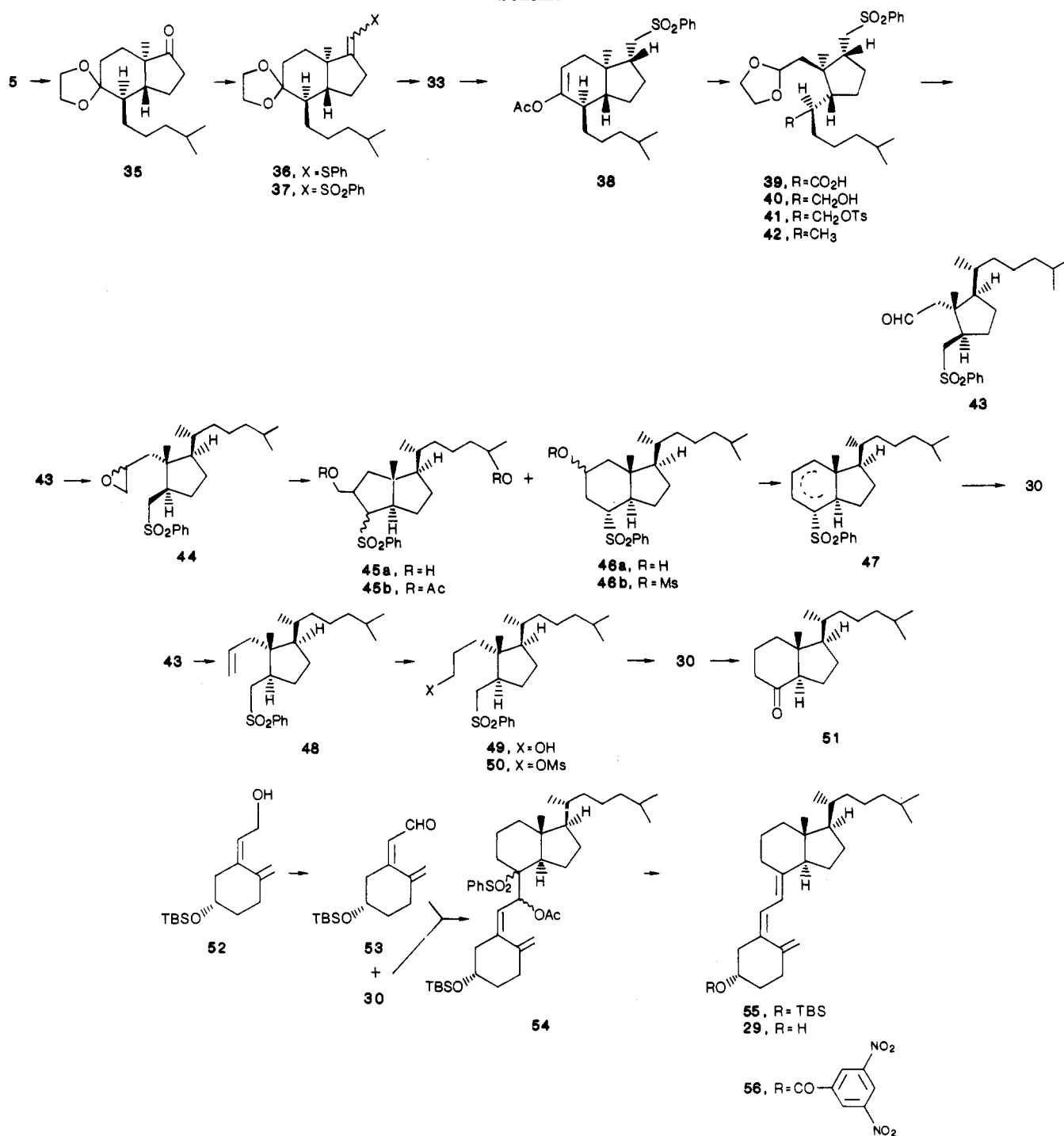
(28) Windaus, A.; Grundmann, W. *Ann.* 1936, 524, 295.

(29) Toh, H. T.; Okamura, W. H. *J. Org. Chem.* 1983, 48, 1414.

(30) Windaus, A.; Schenck, F.; Werder, F. V. *Z. Physiol. Chem.* 1936, 241, 100.

(31) Velluz, L.; Amiard, G.; Petit, A. *Bull. Soc. Chim. Fr.* 1949, 501.

Scheme V



mixture was washed with 10% HCl solution and saturated NaCl solution and dried (Na₂SO₄). Removal of the solvent afforded a crude product which was chromatographed on silica gel (20 g) by using hexane-ethyl acetate (4:1 v/v) as an eluant to give 545 mg (98%) of 10 as a colorless oil: IR (CHCl₃) 1740, 1680 cm⁻¹; NMR (CCl₄) δ 0.87 (3 H, s, CH₃), 5.10 (1 H, d, *J* = 2 Hz, C=CH₂), 5.93 (1 H, d, *J* = 2 Hz, C=CH₂); mass spectrum, *m/z* 178 (M⁺); [α]_D²⁰ -130° (c 0.45). Anal. Calcd for C₁₁H₁₄O₂: C, 74.13; H, 7.92. Found: C, 73.68; H, 7.92.

(-)-(3a*R*,4*R*,7a*R*)-4-(4-Methylpentyl)-3a,4,5,6,7,7a-hexahydro-7a-methylindan-1,5-dione (**5**). To a stirred suspension of CuI (5.13 g, 26.9 mmol) in THF (40 mL) was added dropwise isoamylmagnesium bromide [prepared from Mg (665 mg, 27.4 mmol) and isoamyl bromide (4.06 g, 26.9 mmol)] in THF (40 mL) at -30 °C under an atmosphere of nitrogen. Stirring was continued for 30 min at the same temperature. To this mixture was then added dropwise BF₃·Et₂O (3.3 mL) at -78 °C. After stirring was

continued for 5 min at room temperature, a solution of **10** (4.79 g, 26.9 mmol) in THF (15 mL) was added dropwise and stirred for 10 min at the same temperature. The reaction mixture was quenched with saturated NH₄Cl solution and extracted with ether (200 mL × 3). The combined extracts were washed with water, saturated Na₂S₂O₃, and saturated NaCl solution and dried (Na₂SO₄). The removal of the solvent left a residue, which was chromatographed on silica gel (50 g) by using hexane-ethyl acetate (9:1 v/v) as an eluant to afford 4.9 g (73%) of the diketone **5** as a colorless oil: IR (CHCl₃) 1740, 1710 cm⁻¹; NMR (CDCl₃) δ 0.83 (6 H, d, *J* = 6 Hz, CH(CH₃)₂), 1.12 (3 H, s, CH₃); mass spectrum, *m/z* 250 (M⁺); [α]_D²⁰ -95.8° (c 1.23). Anal. Calcd for C₁₆H₂₆O₂: C, 76.75; H, 10.47. Found: C, 76.53; H, 10.05.

Enol Acetylation of 5. A mixture of **5** (3.37 g, 13.5 mmol), *p*-toluenesulfonic acid (450 mg), and isopropenyl acetate (80 mL) was condensed to 10 mL in 2 h under atmospheric pressure. After refluxing was continued for 4 h, the reaction mixture was diluted

with ethyl acetate (200 mL), washed with ether, saturated NaHCO₃, and NaCl solution, and dried (Na₂SO₄). The residue resulting from the evaporation of the solvent was chromatographed on silica gel (50 g). From the fraction of hexane-ethyl acetate (19:1 v/v), 3.06 g (68%) of (-)-(3*aR*,4*R*,7*aR*)-1,5-diacetoxy-4-(4-methylpentyl)-3*a*,4,7,7*a*-tetrahydro-7*a*-methyl-1-indene (11) was obtained as a colorless oil: IR (CHCl₃) 1755 cm⁻¹; NMR (CDCl₃) δ 0.88 (6 H, d, *J* = 6 Hz, CH(CH₃)₂), 1.00 (3 H, s, CH₃), 2.05 (3 H, s, OCOCH₃), 2.10 (3 H, s, OCOCH₃), 5.25 (1 H, dd, *J* = 3 and 6 Hz, C-6H), 5.55 (1 H, t, *J* = 2 Hz, C-2H); mass spectrum, *m/z* 334 (M⁺); [α]_D²⁰ -35.2° (*c* 0.47); exact mass calcd for C₂₀H₃₀O₄ 334.2142, found 334.2129.

Furthermore, from the fraction of hexane-ethyl acetate (9:1 v/v), 1.03 (26%) of (-)-(3*aR*,4*R*,7*aR*)-5-acetoxy-4-(4-methylpentyl)-3*a*,4,7,7*a*-tetrahydro-7*a*-methylindan-1-one (12) was obtained as a colorless oil: IR (CHCl₃) 1755, 1740 cm⁻¹; NMR (CDCl₃) δ 0.90 (6 H, d, *J* = 6 Hz, CH(CH₃)₂), 1.00 (3 H, s, CH₃), 2.06 (3 H, s, OCOCH₃), 5.30 (1 H, dd, *J* = 3 and 5 Hz, C-6H); mass spectrum, *m/z* 292 (M⁺); [α]_D²⁰ -71.3° (*c* 0.49); exact mass calcd for C₁₈H₂₈O₃ 292.2038, found 292.2033.

Hydrolysis of 11. A mixture of 11 (1.7 g, 5.1 mmol), LiOH (214 mg, 8.9 mmol), water (6 mL), and THF (20 mL) was stirred for 3 h at room temperature. The reaction mixture was diluted with water (20 mL) and extracted with ether (50 mL × 3). The combined extracts were washed with saturated NaCl solution and dried (Na₂SO₄). The residue resulting from the evaporation of the solvent was chromatographed on silica gel (20 g) by using hexane-ethyl acetate (9:1 v/v) as an eluant. Evaporation of the first fraction afforded 1.28 g (86%) of 12, and 130 mg (10%) of 5 was obtained from the second fraction.

(-)-6-Methyl-2(R)-[2-((2,5-dioxacyclopentyl)methyl)-2(R)-methyl-3-oxo-1(R)-cyclopentyl]heptanoic Acid (14). Through a stirred solution of 12 (579 mg, 1.98 mmol) in CH₂Cl₂ (20 mL) was passed O₃ at -78 °C for 30 min. The reaction mixture was then treated with dimethyl sulfide (0.2 g, 3.2 mmol) at the same temperature, washed with water, and dried (Na₂SO₄). The residue resulting from the evaporation of the solvent was dissolved in THF (20 mL) and water (2 mL). To this solution was added LiOH (67 mg, 2.79 mmol), and stirring was continued for 1 h at room temperature. The reaction mixture was then acidified with 1% HCl solution and extracted with ether (50 mL × 3). The combined extracts were washed with saturated NaCl solution and dried (Na₂SO₄). Evaporation of the solvent gave 550 mg of a crude keto acid 13: IR (CHCl₃) 1740, 1730, 1710 cm⁻¹; NMR (CDCl₃) δ 0.88 (6 H, d, *J* = 6 Hz, CH(CH₃)₂), 0.96 (3 H, s, CH₃), 9.00-9.30 (1 H, m, CO₂H), 9.42 (1 H, br s, CHO). This crude 13 was used for the next reaction without further purification because of its instability.

A mixture of the crude 13 (550 mg) obtained above, ethylene glycol (370 mg, 6 mmol), a catalytic amount of camphorsulfonic acid (CSA), and benzene (50 mL) was refluxed in a flask fitted with a Dean-Stark trap. After 1.5 h the reaction mixture was cooled, washed with water, and dried (Na₂SO₄). Evaporation of the solvent gave a crude product which was chromatographed on silica gel (20 g) using hexane-ethyl acetate (3:2 v/v) as an eluant to afford 452 mg (70%) of 14 as a colorless oil: IR (CHCl₃) 1740, 1710 cm⁻¹; NMR (CDCl₃) δ 0.87 (6 H, d, *J* = 6 Hz, CH(CH₃)₂), 0.93 (3 H, s, CH₃), 3.62-4.00 (4 H, m, OCH₂CH₂O), 5.15 (1 H, dd, *J* = 4 and 8 Hz, OCHO), 10.30 (1 H, br s, CO₂H); mass spectrum, *m/z* 326 (M⁺); [α]_D²⁰ -17.6° (*c* 1.3). Anal. Calcd for C₁₈H₃₀O₅: C, 66.23; H, 9.26. Found: C, 65.99; H, 8.88.

(+)-[1(R)-Methyl-2(R)-hydroxy-5(R)-(1(R)-((tosyloxy)methyl)-5-methylhexyl)cyclopentyl]acetaldehyde Ethylene Acetal (16). To a stirred suspension of LiAlH₄ (486 mg, 12.79 mmol) in THF (70 mL) was added dropwise a solution of 14 (1.39 g, 4.3 mmol) in THF (7 mL) at 0 °C, and then stirring was continued for 6 h at room temperature. The reaction mixture was treated successively with aqueous THF (3 mL), 15% NaOH solution (0.5 mL), and then water (1 mL). Evaporation of the filtrate, obtained by passing the reaction mixture through Celite, gave 1.06 g of 15 as a colorless oil: IR (CHCl₃) 3450 cm⁻¹; NMR (CDCl₃) δ 0.87 (3 H, s, CH₃), 0.87 (6 H, d, *J* = 6 Hz, CH(CH₃)₂), 3.63-4.10 (4 H, m, OCH₂CH₂O), 4.97 (1 H, dd, *J* = 7 and 2 Hz, OCHO); mass spectrum, *m/z* 314 (M⁺).

A mixture of crude 15 (1.06 g), Et₃N (1.3 mL), *p*-toluenesulfonyl chloride (700 mg, 3.67 mmol), a catalytic amount of 4-(di-

methylamino)pyridine, and CH₂Cl₂ (50 mL) was stirred for 8 h at room temperature. The reaction mixture was then washed with 10% HCl and saturated NaCl solution and dried (Na₂SO₄). The residue resulting from the evaporation of the solvent was chromatographed on silica gel (20 g) by using hexane-ethyl acetate (3:2 v/v) as an eluant to afford 1.24 g (62%) of 16 as a colorless oil: IR (CHCl₃) 3500 cm⁻¹; NMR (CDCl₃) δ 0.80 (3 H, s, CH₃), 0.84 (6 H, d, *J* = 6 Hz, CH(CH₃)₂), 2.45 (3 H, s, CH₃), 3.70-4.10 (6 H, m, OCH₂CH₂O, CH₂OTs), 4.94 (1 H, dd, *J* = 2 and 7 Hz, OCHO), 7.28 (2 H, d, *J* = 8 Hz, Ar H), 7.73 (2 H, d, *J* = 7 Hz, Ar H); mass spectrum, *m/z* 468 (M⁺); [α]_D²⁰ +21.5° (*c* 0.73). Anal. Calcd for C₂₅H₄₀O₆S: C, 64.07; H, 8.60; S, 6.84. Found: C, 64.45; H, 8.42; S, 7.07.

(+)-[1(R)-Methyl-2(R)-((methoxymethyl)oxy)-5(R)-(1(R)-((tosyloxy)methyl)-5-methylhexyl)cyclopentyl]acetaldehyde Ethylene Acetal (17). To a solution of 16 (626 mg, 1.34 mmol) and *N,N*-diisopropylethylamine (0.47 mL) in CH₂Cl₂ (10 mL) was added dropwise methoxymethyl chloride (0.12 mL) at room temperature, and stirring was continued for 10 h at the same temperature. The reaction mixture was diluted with water (5 mL) and extracted with CH₂Cl₂ (30 mL × 3). The combined extracts were washed with saturated NaCl solution and dried (MgSO₄). Evaporation of the solvent afforded a residue which was chromatographed on silica gel (20 g) by using hexane-ethyl acetate (3:1 v/v) to give 576 mg (84%) of 17 as a colorless oil: NMR (CDCl₃) δ 0.77 (3 H, s, CH₃), 0.85 (6 H, d, *J* = 6 Hz, CH(CH₃)₂), 2.47 (3 H, s, CH₃), 3.30 (3 H, s, OCH₃), 3.60-4.10 (6 H, m, OCH₂CH₂O, CH₂OTs), 4.57 (2 H, s, OCH₂O), 4.88 (1 H, t, *J* = 5 Hz, OCHO), 7.33 (2 H, d, *J* = 8 Hz, Ar H), 7.80 (2 H, d, *J* = 8 Hz, Ar H); [α]_D²⁰ +11.1° (*c* 0.83). Anal. Calcd for C₂₇H₄₄O₇S: C, 63.25; H, 8.65; S, 6.25. Found: C, 62.84; H, 8.42; S, 6.68.

(+)-[1(R)-Methyl-2(R)-((methoxymethyl)oxy)-5(R)-(1(R)-5-dimethylhexyl)cyclopentyl]acetaldehyde Ethylene Acetal (18). To a stirred suspension of LiAlH₄ (50 mg, 1.32 mmol) in THF (25 mL) was added dropwise a solution of 17 (679 mg, 1.28 mmol) in THF (2 mL) at room temperature. The solution was then refluxed for 1 h, after which the reaction mixture was treated successively with aqueous THF (2 mL), 15% NaOH (0.1 mL), and water (0.1 mL). The mixture was then passed through Celite to remove inorganics. The filtrate thus obtained was evaporated to leave a crude product which was chromatographed on silica gel (20 g) by using hexane-ethyl acetate (9:1 v/v), affording 419 mg (92%) of 18 as a colorless oil: NMR (CDCl₃) δ 0.84 (3 H, s, CH₃), 0.89 (6 H, d, *J* = 6 Hz, CH(CH₃)₂), 3.30 (3 H, s, OCH₃), 3.68-4.05 (4 H, m, OCH₂CH₂O), 4.57 (2 H, s, OCH₂O), 4.95 (1 H, t, *J* = 5 Hz, OCHO); mass spectrum, *m/z* 342 (M⁺); [α]_D²⁰ +26.0° (*c* 0.89); exact mass calcd for C₂₀H₃₈O₄ 342.2768, found 342.2742.

(+)-[1(R)-Methyl-2(R)-((methoxymethyl)oxy)-5(R)-(1(R)-5-dimethylhexyl)cyclopentyl]acetaldehyde (19). A solution of 18 (76 mg, 0.22 mmol) in acetone (8 mL) and 10% HCl (3 drops) was stirred for 2 h at room temperature. After evaporation of the solvent, the residue was treated with water (10 mL) and extracted with ether (50 mL × 3). The combined extracts were washed with saturated NaCl solution and dried (Na₂SO₄). Removal of the solvent left a crude product which was chromatographed on silica gel (5 g) by using hexane-ethyl acetate (9:1 v/v) as an eluant to give 61 mg (92%) of 19 as a colorless oil: IR (CHCl₃) 1710 cm⁻¹; NMR (CDCl₃) δ 0.87 (6 H, d, *J* = 6 Hz, CH(CH₃)₂), 0.97 (3 H, s, CH₃), 2.17 (1 H, dd, *J* = 4 and 14 Hz, CH₂CHO), 2.56 (1 H, dd, *J* = 2 and 14 Hz, CH₂CHO), 3.25 (3 H, s, OCH₃), 4.47 (2 H, s, OCH₂O), 9.72 (1 H, dd, *J* = 2 and 4 Hz, CHO); mass spectrum, *m/z* 267 (M⁺ - 31); [α]_D²⁰ +10.9° (*c* 0.64).

2(R)-Methyl-1(R)-[(methoxymethyl)oxy]-2-(3-methoxy-2-propenyl)-3(R)-(1(R)-5-dimethylhexyl)cyclopentane (20). To a solution of (methoxymethyl)diphenylphosphine oxide (273 mg, 1.19 mmol) in THF (7 mL) was added dropwise a solution of LDA [prepared from diisopropylamine (103 mg, 1.02 mmol) and 0.65 mL (1.02 mmol) of *n*-butyllithium (100 mg/mL in hexane)] in THF (2 mL) at 0 °C under stirring. After 10 min, the reaction mixture was cooled to -78 °C, and a solution of 19 (276 mg, 0.93 mmol) in THF (2 mL) was added dropwise to this solution at the same temperature. The resulting mixture was raised to room temperature, quenched with saturated NH₄Cl solution (10 mL), and extracted with ether (50 mL × 3). The

combined extracts were washed with saturated NaCl solution and dried (MgSO_4). To a solution of the residue, resulting from the evaporation of the solvent, in THF (20 mL) was added 93 mg (2.33 mmol) of NaH (60% in oil), and stirring was continued for 20 h at room temperature. The reaction mixture was treated with saturated NH_4Cl solution and extracted with ether (50 mL \times 3). The combined extracts were washed with saturated NaCl solution and dried (MgSO_4). Evaporation of the solvent left a residue which was chromatographed on silica gel (5 g) by using hexane-ethyl acetate (95:5 v/v) as an eluant to give 251 mg (83%) of 20 as a colorless oil: NMR (CDCl_3) δ 0.78 (3 H, s, CH_3), 0.88 (6 H, d, $J = 6$ Hz, $\text{CH}(\text{CH}_3)_2$), 3.27 (3 H, s, OCH_3), 3.49, 3.52 (3 H, each s, OCH_3), 4.48 (2 H, s, OCH_2O), 5.70–6.30 (2 H, m, $\text{CH}=\text{CHO}$); mass spectrum, m/z 326 (M^+), exact mass calcd for $\text{C}_{20}\text{H}_{38}\text{O}_3$ 326.2821, found 326.2843.

(+)-Methyl 3-[1(*R*)-Methyl-2(*R*)-((methoxymethyl)-oxy)-5(*R*)-(1(*R*),5-dimethylhexyl)cyclopentyl]propionate (21). To a stirred solution of 20 (45 mg, 0.14 mmol) in CH_2Cl_2 (2 mL) was added pyridinium chlorochromate (90 mg, 0.42 mmol), and stirring was continued for 20 h at room temperature. The reaction mixture was treated with Florosil and filtered through Celite. The residue resulting from evaporation of the filtrate was chromatographed on silica gel (1 g) by using hexane-ethyl acetate (9:1 v/v) as an eluant to afford 41 mg (87%) of 21 as a colorless oil: IR (CHCl_3) 1730 cm^{-1} ; NMR (CDCl_3) δ 0.82 (3 H, s, CH_3), 0.87 (6 H, d, $J = 6$ Hz, $\text{CH}(\text{CH}_3)_2$), 3.27 (3 H, s, OCH_3), 3.59 (3 H, s, CO_2CH_3), 4.46 (1 H, d, $J = 6$ Hz, OCH_2O), 4.51 (1 H, d, $J = 6$ Hz, OCH_2O); mass spectrum, m/z 311 ($\text{M}^+ - 31$); $[\alpha]_D^{20} +1.3^\circ$ (c 0.62). Anal. Calcd for $\text{C}_{20}\text{H}_{38}\text{O}_4$: C, 70.13; H, 11.18. Found: C, 69.90; H, 10.95.

(+)-Des-AB-8-oxacholestan-9-one (22). A solution of 21 (30 mg, 0.09 mmol) in acetone (5 mL) and 10% HCl solution (3 drops) was refluxed for 4 h. After evaporation of the solvent, the residue was diluted with water (5 mL) and extracted with ether (20 mL \times 3). The combined extracts were washed with saturated NaCl solution and dried (Na_2SO_4). Removal of the solvent afforded the crude product which was chromatographed on silica gel (1 g) by using hexane-ethyl acetate (17:3 v/v) as an eluant to give 20 mg (86%) of 22 as a colorless oil: IR (CHCl_3) 1730 cm^{-1} ; NMR (CDCl_3) δ 0.85 (3 H, s, CH_3), 0.90 (6 H, d, $J = 6$ Hz, $\text{CH}(\text{CH}_3)_2$), 4.20 (1 H, dd, $J = 4$ and 10 Hz, OCH); mass spectrum, m/z 266 (M^+); $[\alpha]_D^{20} +133.2^\circ$ (c 0.62); exact mass calcd for $\text{C}_{17}\text{H}_{30}\text{O}_2$ 266.2246, found 266.2263.

(-)-3-[1(*R*)-Methyl-5(*R*)-(1(*R*),5-dimethylhexyl)-2-oxocyclopentyl]propionic Acid (24). A mixture of 22 (22 mg, 0.08 mmol), LiOH (2.5 mg, 0.10 mmol), water (0.1 mL), and THF (3 mL) was stirred for 1 h at room temperature. The reaction mixture was then acidified with 10% HCl solution and extracted with ethyl acetate (20 mL \times 3). The combined extracts were washed with saturated NaCl solution and dried (Na_2SO_4). The residue resulting from the evaporation of the solvent was dissolved in acetone (3 mL) and Jones' reagent (2 drops) was added to this solution at 0 $^\circ\text{C}$. After 10 min, the reaction mixture was treated with isopropyl alcohol and water successively and extracted with ethyl acetate (20 mL \times 3). The combined extracts were washed with saturated NaCl solution and dried (Na_2SO_4). Evaporation of the solvent left the residue which was chromatographed on silica gel (1 g) by using hexane-ethyl acetate (1:1 v/v) as an eluant to give 20 mg (94%) of 24 as a colorless oil: IR (CHCl_3) 1740, 1710 cm^{-1} ; NMR (CDCl_3) δ 0.88 (6 H, d, $J = 6.6$ Hz, $\text{CH}(\text{CH}_3)_2$), 0.94 (3 H, s, CH_3), 0.99 (3 H, d, $J = 6.6$ Hz, CH_3); mass spectrum, m/z 282 (M^+); $[\alpha]_D^{20} -42.3^\circ$ (c 0.85). Anal. Calcd for $\text{C}_{17}\text{H}_{30}\text{O}_3 \cdot 0.1\text{H}_2\text{O}$: C, 71.84; H, 10.71. Found: C, 71.75; H, 10.52. The IR and NMR spectra of this sample were identical with those of an authentic sample.^{7b}

(-)-Des-AB-8-oxacholest-14-en-9-one (3). A mixture of 24 (47 mg, 0.17 mmol), sodium acetate (13.6 mg, 0.17 mmol), and acetic anhydride (3 mL) was refluxed for 1.5 h. After evaporation of the acetic anhydride, the residue was treated with water and extracted with ether (50 mL \times 3). The combined extracts were washed with saturated NaHCO_3 and NaCl solution successively and then dried (Na_2SO_4). Removal of the solvent afforded the crude product which was chromatographed on silica gel (1 g) by using hexane-ethyl acetate (19:1 v/v) as an eluant to give 35 mg (80%) of 3 as a colorless oil: IR (CHCl_3) 1770 cm^{-1} ; NMR (CDCl_3) δ 0.88 (6 H, d, $J = 6.6$ Hz, $\text{CH}(\text{CH}_3)_2$), 0.93 (3 H, d, $J = 6.3$ Hz,

CH_3), 1.10 (3 H, s, CH_3), 5.02 (1 H, dd, $J = 3.3$ and 1.2 Hz, $\text{C}=\text{CH}$); mass spectrum, m/z 264 (M^+); $[\alpha]_D^{20} -32.5^\circ$ (c 0.63); exact mass calcd for $\text{C}_{17}\text{H}_{30}\text{O}_2$ 264.2087, found 264.2081.

(-)-(3*aR*,4*R*,7*aR*)-5,5-(Ethylenedioxy)-4-(4-methylpentyl)-3*a*,4,5,6,7,7*a*-hexahydro-7*a*-methylindan-1-one (35). A mixture of 5 (2.1 g, 8.4 mmol), ethylene glycol (730 mg, 11.77 mmol), a catalytic amount of *p*-toluenesulfonic acid, and benzene (200 mL) was refluxed in a flask fitted with a Dean-Stark trap. After 1.5 h the reaction mixture was cooled, washed with water, and dried (Na_2SO_4). Evaporation of the solvent gave a crude product which was chromatographed on silica gel (40 g) by using hexane-ethyl acetate (19:1 v/v) as an eluant to afford 2.1 g (85%) of 35 as a colorless oil: IR (CHCl_3) 1740 cm^{-1} ; NMR (CDCl_3) δ 0.87 (6 H, d, $J = 6$ Hz, $\text{CH}(\text{CH}_3)_2$), 0.90 (3 H, s, CH_3), 3.87 (4 H, br s, $\text{OCH}_2\text{CH}_2\text{O}$); mass spectrum, m/z 294 (M^+); $[\alpha]_D^{20} -74.3^\circ$ (c 1.27). Anal. Calcd for $\text{C}_{18}\text{H}_{30}\text{O}_3$: C, 73.43; H, 10.27. Found: C, 73.65; H, 10.46.

(3*aR*,4*R*,7*aR*)-5,5-(Ethylenedioxy)-1-[(phenylthio)methylene]-4-(4-methylpentyl)-3*a*,4,5,6,7,7*a*-hexahydro-7*a*-methylindan (36). After a mixture of 0.47 g (11.75 mmol) of NaH (60% in oil), diethyl [(phenylthio)methyl]phosphonate (2.52 g, 9.7 mmol), and THF (5 mL) was refluxed for 30 min, a solution of the monoketone 35 (1.9 g, 6.5 mmol) in THF (5 mL) was added dropwise to this solution, and refluxing was continued for 5 h. To this end, the reaction mixture was treated with saturated NH_4Cl solution and extracted with ether (50 mL \times 3). The combined extracts were washed with saturated NaCl solution and dried (Na_2SO_4). The residue resulting from the evaporation of the solvent was chromatographed on silica gel (50 g) by using hexane-ethyl acetate (19:1 v/v) as an eluant to give 2.46 g (95%) of 36 as a colorless oil: NMR (CDCl_3) δ 0.87 (6 H, d, $J = 6$ Hz, $\text{CH}(\text{CH}_3)_2$), 0.92, 1.03 (3 H, each s, CH_3), 3.83 (4 H, br s, $\text{OCH}_2\text{CH}_2\text{O}$), 5.57–5.90 (1 H, m, $\text{C}=\text{CHS}$), 6.97–7.43 (5 H, m, Ar H); mass spectrum, m/z 400 (M^+). Anal. Calcd for $\text{C}_{25}\text{H}_{36}\text{O}_2\text{S}$: C, 74.95; H, 9.06; S, 8.00. Found: C, 74.78; H, 8.90; S, 8.27.

(+)-(3*aR*,4*R*,7*aR*)-5,5-(Ethylenedioxy)-1(*Z*)-[(phenylsulfonyl)methylene]-4-(4-methylpentyl)-3*a*,4,5,6,7,7*a*-hexahydro-7*a*-methylindan (37*a*) and (-)-(3*aR*,4*R*,7*aR*)-5,5-(Ethylenedioxy)-1(*E*)-[(phenylsulfonyl)methylene]-4-(4-methylpentyl)-3*a*,4,5,6,7,7*a*-hexahydro-7*a*-methylindan (37*b*). After a mixture of 36 (1 g, 2.5 mmol), saturated NaHCO_3 solution (20 mL), 1.24 g (5.03 mmol) of *m*-chloroperbenzoic acid (70%), and CH_2Cl_2 (20 mL) was stirred for 10 h at room temperature, the reaction mixture was diluted with CH_2Cl_2 (50 mL). The organic layer was washed with water and dried (Na_2SO_4). The residue resulting from the evaporation of the solvent was chromatographed on silica gel (20 g) by using hexane-ethyl acetate (4:1 v/v) as an eluant. Evaporation of the solvent of the first fraction afforded 423 mg (39%) of 37*a* as a colorless oil: NMR (CDCl_3) δ 0.87 (3 H, d, $J = 6$ Hz, $\text{CH}(\text{CH}_3)_2$), 1.17 (3 H, s, CH_3), 3.88 (4 H, br s, $\text{OCH}_2\text{CH}_2\text{O}$), 5.97 (1 H, t, $J = 2$ Hz, $\text{C}=\text{CHSO}_2$), 7.40–7.97 (5 H, m, Ar H); mass spectrum, m/z 432 (M^+); $[\alpha]_D^{20} +25.9^\circ$ (c 1.2). Anal. Calcd for $\text{C}_{25}\text{H}_{36}\text{O}_4\text{S}$: C, 69.41; H, 8.39; S, 7.41. Found: C, 69.77; H, 8.53; S, 7.39.

Furthermore, evaporation of the solvent of the second fraction gave 442 mg (41%) of 37*b* as a colorless oil: NMR (CDCl_3) δ 0.87 (6 H, d, $J = 6$ Hz, $\text{CH}(\text{CH}_3)_2$), 0.90 (3 H, s, CH_3), 2.85 (4 H, br s, $\text{OCH}_2\text{CH}_2\text{O}$), 5.83 (1 H, t, $J = 3$ Hz, $\text{C}=\text{CHSO}_2$), 7.40–7.93 (5 H, m, Ar H); $[\alpha]_D^{20} -78.8^\circ$ (c 1.56). Anal. Calcd for $\text{C}_{25}\text{H}_{36}\text{O}_4\text{S}$: C, 69.41; H, 8.39; S, 7.41. Found: C, 69.46; H, 8.69; S, 7.71.

(-)-(1*S*,3*aR*,4*R*,7*aS*)-1-[(Phenylsulfonyl)methyl]-4-(4-methylpentyl)-3*a*,4,5,6,7,7*a*-hexahydro-7*a*-methylindan-5-one (33). A mixture of 37 (1.3 g, 3 mmol), 5% palladium on carbon (100 mg), 10% HCl solution (0.4 mL), and MeOH (40 mL) was stirred under an atmosphere of hydrogen at room temperature and atmospheric pressure. Hydrogenation was completed within 5 h. The solution was then filtered to remove the catalyst. The residue resulting from the evaporation of the filtrate was dissolved in benzene. The benzene layer was washed with saturated NaCl solution and dried (Na_2SO_4). Removal of the solvent yielded a crude product, which was chromatographed on silica gel (20 g) by using hexane-ethyl acetate (17:3 v/v) as an eluant to give 845 mg (72%) of 33 as a colorless oil: IR (CHCl_3) 1710 cm^{-1} ; NMR (CDCl_3) δ 0.87 (6 H, d, $J = 6$ Hz, $\text{CH}(\text{CH}_3)_2$), 0.88 (3 H, s, CH_3), 2.87–3.13 (2 H, m, CH_2SO_2), 7.40–7.97 (5 H, m, Ar H); mass spectrum, m/z 390 (M^+); $[\alpha]_D^{20} -26.0^\circ$ (c 1.49). Anal. Calcd for

$C_{23}H_{34}O_3S$: C, 70.73; H, 8.77; S, 8.21. Found: C, 70.45; H, 8.95; S, 8.67.

(-)-**(1*S*,3*aR*,4*R*,7*aS*)-5-Acetoxy-1-[(phenylsulfonyl)methyl]-4-(4-methylpentyl)-3*a*,4,7,7*a*-tetrahydro-7*a*-methylindan (38)**. A mixture of **33** (3.62 g, 9.3 mmol), *p*-toluenesulfonic acid (500 mg), and isopropenyl acetate (70 mL) was condensed to 10 mL in 2 h under an atmospheric pressure. After refluxing was continued for 10 h, the reaction mixture was diluted with ethyl acetate (200 mL), washed with water, saturated $NaHCO_3$, and NaCl solution, and dried (Na_2SO_4). The residue resulting from the evaporation of the solvent was chromatographed on silica gel (50 g) by using hexane-ethyl acetate (17:3 v/v) as an eluant to give 3.94 g (98%) of **38** as a colorless oil: IR ($CHCl_3$) 1740 cm^{-1} ; NMR ($CDCl_3$) δ 0.73 (3 H, s, CH_3), 0.85 (6 H, d, $J = 6\text{ Hz}$, $CH(CH_3)_2$), 2.10 (3 H, s, CH_3), 5.10–5.37 (1 H, m, C=CH), 7.45–8.00 (5 H, m, Ar H); $[\alpha]_D^{20} -0.38^\circ$ (c 2.1). Anal. Calcd for $C_{25}H_{36}O_4S$: C, 69.41; H, 8.39; S, 7.41. Found: C, 69.07; H, 8.34; S, 7.74.

(+)-**6-Methyl-2(R)-[2-((2,5-dioxacyclopentyl)methyl)-2(S)-methyl-3(S)-((phenylsulfonyl)methyl)-1(R)-cyclopentyl]heptanoic Acid (39)**. Through a stirred solution of **38** (787 mg, 1.82 mmol) in CH_2Cl_2 (30 mL) was passed O_3 at -78°C for 30 min. To this end, the reaction mixture was treated with dimethyl sulfide (0.16 mL) at the same temperature, washed with water, and dried (Na_2SO_4). The residue resulting from the evaporation of the solvent was dissolved in THF (20 mL) and water (3 mL). To this solution was added LiOH (76 mg, 3.17 mmol), and stirring was continued for 2 h at room temperature. The reaction mixture was then acidified with 10% HCl solution and extracted with CH_2Cl_2 (50 mL \times 3). The combined extracts were washed with saturated NaCl solution and dried ($MgSO_4$). Evaporation of the solvent gave 750 mg of the crude acid aldehyde: IR ($CHCl_3$) $1720, 1700\text{ cm}^{-1}$; NMR ($CDCl_3$) δ 0.73 (3 H, s, CH_3), 0.83 (6 H, d, $J = 6\text{ Hz}$, $CH(CH_3)_2$), 7.40–8.00 (5 H, m, Ar H), 8.33–8.82 (1 H, m, CO_2H), 9.47 (1 H, t, $J = 2.5\text{ Hz}$, CHO). This crude sample was used for the next reaction without further purification because of its instability.

A mixture of the crude acid aldehyde (750 mg) obtained above, ethylene glycol (170 mg, 2.74 mmol), a catalytic amount of camphorsulfonic acid, and benzene (50 mL) was refluxed in a flask fitted with a Dean-Stark trap. After 2 h the reaction mixture was cooled, washed with water, and dried (Na_2SO_4). Evaporation of the solvent gave a crude product which was chromatographed on silica gel (20 g) by using hexane-ethyl acetate (7:3 v/v) as an eluant to afford 620 mg (73%) of **39** as a colorless oil: IR ($CHCl_3$) 1710 cm^{-1} ; NMR ($CDCl_3$) δ 0.70 (3 H, s, CH_3), 0.83 (6 H, d, $J = 6\text{ Hz}$, $CH(CH_3)_2$), 3.45–3.83 (4 H, m, OCH_2CH_2O), 4.70 (1 H, dd, $J = 3$ and 7 Hz , OCHO), 7.40–8.03 (5 H, m, Ar H), 8.50–8.87 (1 H, m, CO_2H); mass spectrum, m/z 466 (M^+); $[\alpha]_D^{20} +0.79^\circ$ (c 1.5); exact mass calcd for $C_{25}H_{38}O_6S$ 466.2390, found 466.2417.

(+)-**[1(S)-Methyl-2(S)-((phenylsulfonyl)methyl)-5(R)-1(S)-(hydroxymethyl)-5-methylhexyl)cyclopentyl]acetaldehyde Ethylene Acetal (40)**. To a stirred suspension of $LiAlH_4$ (122 mg, 3.2 mmol) in THF (20 mL) was added dropwise a solution of **39** (619 mg, 1.33 mmol) in THF (3 mL) at 0°C , and then stirring was continued for 9 h at room temperature. The reaction mixture was treated successively with aqueous THF (2 mL), 15% NaOH solution (0.2 mL), and then water (1 mL). Evaporation of the filtrate, obtained by passing the reaction mixture through Celite, gave 435 mg (72%) of **40** as a colorless oil: IR ($CHCl_3$) 3460 cm^{-1} ; NMR ($CDCl_3$) δ 0.63 (3 H, s, CH_3), 0.85 (6 H, d, $J = 6\text{ Hz}$, $CH(CH_3)_2$), 3.24–3.87 (6 H, m, OCH_2CH_2O , CH_2O), 4.55 (1 H, t, $J = 4\text{ Hz}$, OCHO), 7.37–7.97 (5 H, m, Ar H); mass spectrum, m/z 452 (M^+); $[\alpha]_D^{20} +0.88^\circ$ (c 2.0). Anal. Calcd for $C_{25}H_{40}O_5S$: C, 66.33; H, 8.91; S, 7.08. Found: C, 66.61; H, 9.13; S, 7.17.

(-)-**[1(S)-Methyl-2(S)-((phenylsulfonyl)methyl)-5(R)-1(S)-((tosyloxy)methyl)-5-methylhexyl)cyclopentyl]acetaldehyde Ethylene Acetal (41)**. A mixture of **40** (190 mg, 0.42 mmol), *p*-toluenesulfonyl chloride (100 mg, 0.52 mmol), a catalytic amount of 4-(dimethylamino)pyridine, and pyridine (1 mL) was stirred for 5 h at room temperature. The reaction mixture was then diluted with water (5 mL) and extracted with ether (50 mL \times 3). The combined extracts were washed with 10% HCl solution and saturated NaCl solution, and dried (Na_2SO_4). The residue resulting from the evaporation of the solvent was chromatographed

on silica gel (5 g) by using hexane-ethyl acetate (7:3 v/v) as an eluant to give 237 mg (93%) of **41** as a colorless oil: NMR ($CDCl_3$) δ 0.58 (3 H, s, CH_3), 0.80 (6 H, d, $J = 6\text{ Hz}$, $CH(CH_3)_2$), 2.42 (3 H, s, CH_3), 3.52–3.83 (4 H, m, OCH_2CH_2O), 3.92 (2 H, d, $J = 4\text{ Hz}$, CH_2O), 4.52 (1 H, t, $J = 5\text{ Hz}$, OCHO), 7.23 (2 H, d, $J = 8\text{ Hz}$, Ar H), 7.68 (2 H, d, $J = 8\text{ Hz}$, Ar H), 7.42–7.95 (5 H, m, Ar H); mass spectrum, m/z 606 (M^+); $[\alpha]_D^{20} -3.1^\circ$ (c 2.0). Anal. Calcd for $C_{32}H_{46}O_7S_2$: C, 63.33; H, 7.64; S, 10.57. Found: C, 63.33; H, 7.43; S, 10.64.

(-)-**[1(S)-Methyl-2(S)-((phenylsulfonyl)methyl)-5(R)-1(R)-5-dimethylhexyl)cyclopentyl]acetaldehyde Ethylene Acetal (42)**. To a stirred suspension of $LiAlH_4$ (25 mg, 0.67 mmol) in THF (15 mL) was added dropwise a solution of **41** (405 mg, 0.67 mmol) in THF (3 mL) at room temperature. After stirring had been continued for 3 h under refluxing, the reaction mixture was treated successively with aqueous THF (1 mL), 15% NaOH solution (0.05 mL), and water (0.05 mL). The mixture was then passed through Celite to remove inorganics. The filtrate thus obtained was evaporated to leave a crude product, which was chromatographed on silica gel (10 g) by using hexane-ethyl acetate (17:3 v/v) as an eluant to give 276 mg (95%) of **42** as a colorless oil: NMR ($CDCl_3$) δ 0.67 (3 H, s, CH_3), 0.85 (6 H, d, $J = 6\text{ Hz}$, $CH(CH_3)_2$), 3.50–3.80 (4 H, m, OCH_2CH_2O), 4.67 (1 H, t, $J = 4.5\text{ Hz}$, OCHO), 7.77–8.00 (5 H, m, Ar H); mass spectrum, m/z 436 (M^+); $[\alpha]_D^{20} -12.0^\circ$ (c 2.4). Anal. Calcd for $C_{25}H_{40}O_4S$: C, 68.76; H, 9.23; S, 7.34. Found: C, 68.44; H, 9.02; S, 7.71.

(-)-**[1(S)-Methyl-2(S)-((phenylsulfonyl)methyl)-5(R)-1(R)-5-dimethylhexyl)cyclopentyl]acetaldehyde (43)**. A solution of **42** (306 mg, 0.7 mmol) in acetone (15 mL) and 10% HCl solution (1.5 mL) was stirred for 6 h at room temperature. After evaporation of the solvent, the residue was treated with water (10 mL) and extracted with ether (50 mL \times 3). The combined extracts were washed with saturated NaCl solution and dried (Na_2SO_4). Removal of the solvent left a crude product, which was chromatographed on silica gel (5 g) by using hexane-ethyl acetate (9:1 v/v) as an eluant to give 261 mg (95%) of **43** as a colorless oil: IR ($CHCl_3$) 1720 cm^{-1} ; NMR ($CDCl_3$) δ 0.73 (3 H, s, CH_3), 0.87 (6 H, d, $J = 6\text{ Hz}$, $CH(CH_3)_2$), 7.50–8.03 (5 H, m, Ar H), 9.63 (1 H, t, $J = 3\text{ Hz}$, CHO); mass spectrum, m/z 392 (M^+); $[\alpha]_D^{20} -19.3^\circ$ (c 1.54); exact mass calcd for $C_{23}H_{36}O_3S$ 392.2383, found 392.2347.

2(S)-Methyl-3(R)-1(R)-5-dimethylhexyl-1(S)-[(phenylsulfonyl)methyl]-2-(2,3-epoxypropyl)cyclopentane (44). To a stirred suspension of trimethylsulfonium iodide (39 mg, 0.19 mmol) in THF (1 mL) was added dropwise 0.135 mL (0.21 mmol) of *n*-butyllithium (100 mg/mL in hexane) at 0°C . After 10 min, a solution of **43** (31 mg, 0.08 mmol) in THF (0.5 mL) was added to this solution, and stirring was continued for 30 min at the same temperature. The reaction mixture was then stirred for 1 h at room temperature and evaporated to leave a residue, which was treated with water (5 mL) and extracted with ether (20 mL \times 3). The combined extracts were washed with water and dried (Na_2SO_4). Evaporation of the solvent afforded a crude product, which was chromatographed on silica gel (1 g) by using hexane-ethyl acetate (9:1 v/v) as an eluant to give 10 mg (31%) of **44** as a colorless oil: NMR ($CDCl_3$) δ 0.67 (3 H, s, CH_3), 0.84 (6 H, d, $J = 6\text{ Hz}$, $CH(CH_3)_2$), 2.29 (1 H, dd, $J = 5$ and 2.5 Hz , \overline{OCCHH}), 2.65 (1 H, dd, $J = 5$ and 4 Hz , \overline{OCCHH}), 2.66–2.94 (1 H, m, \overline{OCHC}), 7.48–7.98 (5 H, m, Ar H); mass spectrum, m/z 265 ($M^+ - 141$). Anal. Calcd for $C_{24}H_{38}O_3S \cdot 0.2H_2O$: C, 70.27; H, 9.44; S, 7.82. Found: C, 70.00; H, 9.36; S, 8.08.

Intramolecular Cyclization of 44. To a stirred solution of LDA [prepared from diisopropylamine (13 mg, 0.13 mmol) and 0.08 mL (0.13 mmol) of *n*-butyllithium (100 mg/mL in hexane)] in THF (2 mL) was added a solution of **44** (26 mg, 0.06 mmol) in THF (0.5 mL) at -78°C , and stirring was continued for 10 min at the same temperature. The reaction mixture was treated with saturated NH_4Cl solution and extracted with ether (20 mL \times 3). The combined extracts were washed with saturated NaCl solution and dried (Na_2SO_4). Evaporation of the solvent yielded a crude product which was chromatographed on silica gel (1 g). From the fraction eluted with hexane-ethyl acetate (9:1 v/v), 8 mg (30%) of **45a** was obtained as a colorless oil: IR ($CHCl_3$) 3500 cm^{-1} ; NMR ($CDCl_3$) δ 0.72 (3 H, s, CH_3), 0.83 (6 H, d, $J = 6\text{ Hz}$, $CH(CH_3)_2$),

3.04–3.36 (1 H, m, CHSO₂), 3.55–3.76 (2 H, m, CH₂O), 7.44–7.94 (5 H, m, Ar H); mass spectrum, m/z 406 (M⁺); exact mass calcd for C₂₄H₃₈O₃S 406.2542, found 406.2542.

Furthermore, from the fraction eluted with hexane–ethyl acetate (17:3, v/v), 13 mg (50%) of **45b** was obtained as a colorless oil: NMR (CDCl₃) δ 0.72 (3 H, s, CH₃), 0.85 (6 H, d, $J = 6$ Hz, CH(CH₃)₂), 3.04 (1 H, ddd, $J = 12, 12,$ and 4 Hz, CHSO₂), 3.60–4.00 (1 H, m, CHO), 7.46–7.92 (5 H, m, Ar H); mass spectrum, m/z 265 (M⁺ – 141); exact mass calcd for C₁₈H₃₃O 265.2531 (M⁺ – 141), found 265.2538.

Acetylation of 45a. After a mixture of **45a** (2 mg, 0.005 mmol), acetic anhydride (2 drops), and pyridine (0.05 mL) had been stirred for 2 h at room temperature, the reaction mixture was diluted with water (5 mL) and extracted with ether (5 mL \times 3). The combined extracts were washed with saturated NaCl solution and dried (Na₂SO₄). The residue resulting from the evaporation of the solvent was chromatographed on silica gel (1 g) by using hexane–ethyl acetate (9:1 v/v) as an eluant to give 2 mg (91%) of **45b** as a colorless oil: IR (CHCl₃) 1730 cm⁻¹; NMR (CDCl₃) δ 0.72 (3 H, s, CH₃), 0.82 (6 H, d, $J = 6$ Hz, CH(CH₃)₂), 2.96–3.32 (1 H, m, CHSO₂), 3.80–4.20 (2 H, m, CH₂O), 7.40–7.92 (5 H, m, Ar H); mass spectrum, m/z 448 (M⁺); exact mass calcd for C₂₆H₄₀O₄S 448.2647, found 448.2657.

(+)-8 α -(Phenylsulfonyl)-des-AB-cholestane (30). After a mixture of **45a** (3 mg, 0.08 mmol), mesyl chloride (1 drop), and pyridine (0.1 mL) was stirred for 1 h at 0 °C, the reaction mixture was diluted with water (5 mL) and extracted with ether (5 mL \times 3). The combined extracts were washed with 10% HCl and saturated NaCl solution, and dried (Na₂SO₄). Evaporation of the solvent afforded a crude product, **46b**, which was used for the next reaction without further purification.

A mixture of the crude mesylate **46b** obtained above, LiBr (1 mg, 0.01 mmol), Li₂CO₃ (1 mg, 0.01 mmol), and *N,N*-dimethylformamide (0.5 mL) was stirred for 4 h at 150 °C. The reaction mixture was then diluted with water and extracted with ethyl acetate (5 mL \times 3). The combined extracts were washed with water and dried (Na₂SO₄). Removal of the solvent yielded **47** as a crude product: NMR (CDCl₃) δ 0.68, 0.78 (3 H, each s, CH₃), 0.85 (6 H, d, $J = 6$ Hz, CH(CH₃)₂), 3.20–3.68 (1 H, m, CHSO₂), 5.20–6.23 (2 H, m, CH=CH), 7.40–7.92 (5 H, m, Ar H); mass spectrum m/z 247 (M⁺ – 141).

Next, a mixture of crude **47** prepared above and 5% palladium on carbon (1 mg) in ethyl acetate (1 mL) was stirred under an atmosphere of hydrogen at room temperature for 8 h. The solution was then filtered to remove the catalyst, and the filtrate was evaporated to leave a residue which was chromatographed on silica gel (1 g) by using hexane–ethyl acetate (9:1 v/v) as an eluant. Evaporation of the solvent yielded 2 mg (69%) of **30** as colorless needles: mp 116.5–118 °C (petroleum ether); NMR (CDCl₃) δ 0.69 (3 H, s, CH₃), 0.85 (6 H, d, $J = 6$ Hz, CH(CH₃)₂), 2.84–3.16 (1 H, m, CHSO₂), 7.44–7.92 (5 H, m, Ar H); mass spectrum, m/z 249 (M⁺ – 141); $[\alpha]_D^{20} +0.06^\circ$ (c 0.2). Anal. Calcd for C₂₄H₃₈O₂S: C, 73.39; H, 9.81; S, 8.21. Found: C, 73.62; H, 9.91; S, 8.43.

(-)-2(S)-Methyl-1(S)-[(phenylsulfonyl)methyl]-2-(2-propenyl)-3(R)-(1(R),5-dimethylhexyl)cyclopentane (48). To a solution of [(trimethylsilyl)methyl]magnesium chloride [prepared from 36 mg (1.48 mmol) of Mg and 182 mg (1.48 mmol) of (chloromethyl)trimethylsilane] in ether (8 mL) was added dropwise a solution of **43** (224 mg, 0.57 mmol) in ether (1 mL) at room temperature, and stirring was continued for 1 h at the same temperature. The reaction mixture was poured into saturated NH₄Cl solution, and extracted with ether (20 mL \times 3). The combined extracts were washed with saturated NaHCO₃ and NaCl solution and dried (Na₂SO₄). The residue resulting from the evaporation of the solvent was dissolved in THF (20 mL). To this solution was added 24 mg (0.6 mmol) of NaH (60% in oil) under stirring. After refluxing had been continued for 12 h, the reaction mixture was poured into saturated NH₄Cl solution and extracted with ether (20 mL \times 3). The combined extracts were washed with saturated NaCl solution and dried (Na₂SO₄). Evaporation of the solvent gave the residue, which was chromatographed on silica gel (5 g) by using hexane–ethyl acetate (19:1 v/v) as an eluant to afford 188 mg (81%) of **48** as a colorless oil: NMR (CDCl₃) δ 0.63 (3 H, s, CH₃), 0.85 (6 H, d, $J = 6$ Hz, CH(CH₃)₂), 4.70 (1 H, dd, $J = 16$ and 2 Hz, CH=CH₂), 4.83 (1 H, dd, $J = 8$ and 2 Hz, CHCH₂), 5.30–5.80 (1 H, m, CH=CH₂),

7.20–7.92 (5 H, m, Ar H); mass spectrum, m/z 390 (M⁺); $[\alpha]_D^{20} -12.6^\circ$ (c 1.4); exact mass calcd for C₂₄H₃₈O₂S 390.2593, found 390.2593.

(-)-2(S)-Methyl-2-(3-hydroxypropyl)-1(S)-[(phenylsulfonyl)methyl]-3(R)-(1(R),5-dimethylhexyl)cyclopentane (49). To a stirred solution of **48** (159 mg, 0.4 mmol) in hexane (8 mL) was added BH₃·SMe₂ (0.013 mL) at 0 °C, and stirring was continued for 30 min at the same temperature. After stirring was continued for 1 h at room temperature, EtOH (8 mL) and 10% NaOH solution (0.054 mL) were added to this solution. The reaction mixture was cooled to 0 °C and then treated with 30% H₂O₂ solution (0.046 mL). After the mixture was stirred for 30 min at 0 °C, stirring was continued for 1 h at 50 °C. Finally, the reaction mixture was poured into ice–water (80 mL) and extracted with ether (20 mL \times 3). The combined extracts were washed with saturated NaCl solution and dried (K₂CO₃). The residue resulting from the removal of the solvent was chromatographed on silica gel (5 g) by using hexane–ethyl acetate (7:3 v/v) as an eluant to give 153 mg (92%) of **49** as a colorless oil: IR (CHCl₃) 3500 cm⁻¹; NMR (CDCl₃) δ 0.63 (3 H, s, CH₃), 0.87 (6 H, d, $J = 6$ Hz, CH(CH₃)₂), 3.37–3.70 (2 H, m, CH₂O), 7.48–8.07 (5 H, m, Ar H); mass spectrum, m/z 408 (M⁺); $[\alpha]_D^{20} -4.8^\circ$ (c 1.17); exact mass calcd for C₂₄H₄₀O₃S 408.2697, found 408.2667.

(-)-2(S)-Methyl-2-[3-(mesyloxy)propyl]-1(S)-[(phenylsulfonyl)methyl]-3(R)-(1(R),5-dimethylhexyl)cyclopentane (50). After a mixture of **49** (144 mg, 0.35 mmol), mesyl chloride (0.027 mL), and pyridine (1 mL) had been stirred for 1 h at 0 °C, the reaction mixture was diluted with water (10 mL) and extracted with ether (20 mL \times 3). The combined extracts were washed with 10% HCl and saturated NaCl solution, and dried (Na₂SO₄). Evaporation of the solvent afforded a residue which was chromatographed on silica gel (5 g) by using hexane–ethyl acetate (7:3 v/v) as an eluant to give 168 mg (98%) of **50** as a colorless oil: NMR (CDCl₃) δ 0.63 (3 H, s, CH₃), 0.85 (6 H, d, $J = 6$ Hz, CH(CH₃)₂), 3.00 (3 H, s, OSO₂CH₃), 3.97–4.23 (2 H, m, CH₂O), 7.47–8.00 (5 H, m, Ar H); mass spectrum, m/z 390 (M⁺ – 96); $[\alpha]_D^{20} -0.42^\circ$ (c 0.96). Anal. Calcd for C₂₅H₄₂O₅S₂: C, 61.69; H, 8.70; S, 13.18. Found: C, 62.07; H, 8.98; S, 13.32.

Intramolecular Cyclization of 50. To a stirred solution of LDA [prepared from diisopropylamine (39 mg, 0.39 mmol) and 0.25 mL (0.39 mmol) of *n*-butyllithium (100 mg/mL in hexane)] in THF (10 mL) was added dropwise a solution of **50** (155 mg, 0.32 mmol) in THF (1 mL) at –78 °C, and stirring was continued for 30 min at the same temperature. After stirring had been continued for 30 min at room temperature, the reaction mixture was treated with saturated NH₄Cl solution and extracted with ether (20 mL \times 3). The combined extracts were washed with saturated NaCl solution and dried (Na₂SO₄). The residue resulting from removal of the solvent was chromatographed on silica gel (5 g) by using hexane–ethyl acetate (9:1 v/v) as an eluant to give 105 mg (84%) of **30** as colorless needles: mp 116.5–118 °C, $[\alpha]_D^{20} +0.06^\circ$ (c 1.4). This was identical with the sample prepared above.

Grundmann's Ketone (51). To a solution of **30** (10 mg, 0.026 mmol) in THF (3 mL) was added a solution of LDA [prepared from diisopropylamine (8.4 mg, 0.083 mmol) and 0.05 mL (0.08 mmol) of *n*-butyllithium (100 mg/mL in hexane)] in THF (1 mL) at –78 °C. After the mixture was stirred at the same temperature for 10 min, a solution of MoO₅·HMPA·pyridine²⁷ (37.8 mg, 0.09 mmol) in THF (1 mL) was added under stirring. The mixture was stirred for 20 min at –78 °C and then quenched with saturated Na₂SO₃ solution. The reaction mixture was poured into water (5 mL) and extracted with ether (10 mL \times 3). The combined extracts were washed with 10% HCl and saturated NaCl solution and dried (MgSO₄). Evaporation of the solvent afforded a crude product, which was chromatographed on silica gel (1 g) by using hexane–ethyl acetate (10:1 v/v) as an eluant to give 4.2 mg (62%) of **51** as a colorless oil. This was identical with an authentic sample²⁸ in all aspects.

Vitamin D₃ (29). To a solution of **52** (50 mg, 0.18 mmol) in THF (10 mL) was added MnO₂ (500 mg, 3.5 mmol), and stirring was continued for 6 h at room temperature. After filtration of inorganics, the filtrate was evaporated to leave a residue, which was chromatographed on silica gel (2 g) by using hexane–ethyl acetate (19:1 v/v) as an eluant to give 39 mg of **53** as a colorless oil, which was used immediately for the next reaction because of its instability.

A solution of LDA [prepared from diisopropylamine (16 mg, 0.16 mmol) and 0.1 mL (0.16 mmol) of *n*-butyllithium (100 mg/mL in hexane)] in THF (1 mL) was added dropwise to a stirred solution of **30** (58 mg, 0.149 mmol) in THF (1 mL) at -78°C . After stirring had been continued for 20 min, a solution of **53** (39 mg, 0.15 mmol) in THF (0.5 mL) was added to this solution, and the mixture was stirred for 1 h at the same temperature. The reaction mixture was then treated with acetyl chloride (0.011 mL), poured into water, and extracted with ether (10 mL \times 3). The combined extracts were washed with saturated NaCl solution and dried (Na_2SO_4). The residue resulting from the evaporation of the solvent was chromatographed on silica gel (1 g) by using hexane-ethyl acetate (97:3 v/v) as an eluant to give 85 mg of **54** as a colorless oil, which was used immediately for the next reaction.

To a stirred solution of **54** (85 mg) in THF-MeOH (1:1) (3 mL) was added 5% sodium amalgam (300 mg) at -20°C , and stirring was continued for 2 h at the same temperature. After stirring had been continued for 7 h at room temperature, the reaction mixture was poured into water and extracted with ether (20 mL \times 3). The combined extracts were washed with saturated NaCl solution and dried (Na_2SO_4). The crude product, **55**, obtained by the evaporation of the solvent was dissolved in THF (2 mL), treated with 0.13 mL of *n*-Bu₄NF (1.0 M in THF), and stirred for 2 h at room temperature. The mixture was then poured into water and extracted with ether (20 mL \times 3). The combined

extracts were washed with saturated NaCl solution and dried (Na_2SO_4). The residue resulted from the evaporation of the solvent was chromatographed on silica gel (1 g) by using hexane-ethyl acetate (17:3 v/v) as an eluant to give 29.2 mg (51%) of vitamin D₃ (**29**) as a colorless oil: $[\alpha]_{\text{D}}^{20} +49.5^{\circ}$ (*c* 0.5) [authentic sample, $[\alpha]_{\text{D}}^{20} +51.2^{\circ}$ (*c* 0.63)]. This sample was identical with an authentic sample in spectral comparisons.

3,5-Dinitrobenzoate of Vitamin D₃. To a solution of triethylamine (0.08 mL) and 3,5-dinitrobenzoyl chloride (10 mg, 0.04 mmol) in CH_2Cl_2 (2 mL) was added a solution of **29** (5 mg, 0.013 mmol) and stirring was continued for 1 h at room temperature. The reaction mixture was then poured into water and extracted with ether (10 mL \times 3). The combined extracts were washed with saturated NaCl solution and dried (Na_2SO_4). Evaporation of the solvent gave a crude product, which was chromatographed on neutral alumina (1 g) by using hexane-ethyl acetate (19:1 v/v) as an eluant to give 6 mg (80%) of **56** as yellow needles: mp $129-130^{\circ}\text{C}$ (acetone-MeOH); $[\alpha]_{\text{D}}^{20} +95.9^{\circ}$ (*c* 0.34). This was identical with the authentic sample^{30,31} in all aspects.

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α -Amino Acid Derivatives as Chiral Educts for Asymmetric Products. Synthesis of Sphingosine from α' -Amino- α,β -ynones

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The utility of α' -amino- α,β -ynones in the chiroselective synthesis of sphingosine is demonstrated. Thus, a protected L-serine isoxazolidide has been converted to sphingosine by two routes, both via α,β -ynones. The first route is very short and high yielding, merely involving two selective reductions after synthesis of the appropriate α,β -ynone. The second route involves alkylation of a β -unsubstituted ynone and illustrates the synthetic versatility of the α' -amino- α,β -ynone system. Further routes through conjugate 1,4 additions to ynones are demonstrated but are limited by the highly reactive nature of this system.

The chiroselective synthesis of natural products from simple, readily available components of the chiral pool as educts has become a major objective for organic chemists. Previous work from this laboratory has demonstrated the conversion of suitably protected amino acids into optically pure amino ketones by acylation of aromatic rings¹ and by reaction with organometallic reagents.² A modification of this methodology, using isoxazolidides, has produced optically pure α' -amino- α,β -ynones.³ This modification was necessary because the less reactive organometallics derived from acetylenic precursors would not react with the carboxylate functionality of protected amino acids. The use of an isoxazolidide "activating" group gave good yields of the desired α,β -ynones.

To demonstrate the α' -amino- α,β -ynone system's utility, we have synthesized the amino lipid sphingosine (1).

Presented here are two chiroselective routes to sphingosine from L-serine. Both routes involve the use of α' -amino- α,β -ynones and exploit the flexibility inherent in these synthetic intermediates. The carbon chain can be introduced directly via lithium pentadecyne, followed by a diastereoselective reduction of the ketone. This produces sphingosine (1) in five steps from CBZ-serine in an overall yield of 22% (Scheme I and II). An alternative path is to form an α,β -unsubstituted acetylenic ketone (17; Scheme III) using lithium acetylide. Reduction of the ketone followed by alkylation with 1-iodotridecane (Scheme IV) then leads to sphingosine (1).

Sphingosine has been the target of several syntheses in the past.⁴ Recently, interest has greatly increased, with methodology centering on enantiospecific routes. Attempts have been made at utilizing the 2*S* stereochemistry of L-serine as a chiral precursor;⁵ however, this route involves an intermediate α -amino aldehyde, whose chiral integrity is questionable.⁶ Sphingosine has also been synthesized

(1) Buckley, T. F., III; Rapoport, H. *J. Am. Chem. Soc.* **1981**, *103*, 6157-6163.

(2) (a) Kundsén, C. G.; Rapoport, H., *J. Org. Chem.* **1983**, *48*, 2260-2266. (b) Maurer, P. J.; Takahata, H.; Rapoport, H. *J. Am. Chem. Soc.* **1984**, *106*, 1095-1098. (c) Maurer, P. J.; Knudsen, C. G.; Palkowitz, A. D.; Rapoport, H. *J. Org. Chem.* **1985**, *50*, 325-330.

(3) Cupps, T. L.; Boutin, R. H.; Rapoport, H. *J. Org. Chem.* **1985**, *50*, 3972-3979.

(4) Syntheses of sphingosine are reviewed to 1967 by: Shapiro, D. In *Chemistry of Sphingolipids*; Hermann: Paris, 1969.

(5) (a) Newman, H. *J. Am. Chem. Soc.* **1973**, *95*, 4098-4099. (b) Tkaczuk, P.; Thornton, E. R. *J. Org. Chem.* **1981**, *46*, 4393-4398.